

## **Physico-Chemical Characterization of farmland Soil used in some villages of Radhanpur Taluka. Dist :Patan (Gujarat) India**

**Chirag R. Chaudhary, Dinesh S. Chaudhari, Upendra R. Patel & Pradhuman A. Parmar\***

Department of Chemistry, M.N. College Visnagar, Gujarat, India.

### **Abstract:**

This Physico-Chemical study of soil is based on various parameter like pH, Electrical Conductivity (EC), Total Organic Carbon, Organic Carbon(OC), Available Phosphorus ( $P_2O_5$ ) and available Potassium ( $K_2O$ ). Five representative locations were selected for the study and 20 samples from each location and direction of area were collected. This study leads us to the conclusion of the nutrient's quantity of soil of Radhanpur Taluka, Dist. Patan, Gujarat State. Soil sampling is the most vital step for any soil Analysis. As a very small fraction of the huge soil mass is used for analysis. It becomes extremely important to get a truly representative soil sample of the field. Results show that most of parameters are in medium range. This information will help farmers to decide the problems related to soil nutrients amount of fertilizers to be added to soil to make production economic.

Key Words: Quality of soil, EC, pH.

### **Introduction:**

Soils are composed of solids (minerals and organic matter), and pores which hold water and air. The bulk density of sample is known as weight of mass water sample divided by the bulk volume. Bulk density is influenced by the amount of organic matter in soils, their texture, constituent minerals and porosity. Knowledge of soil bulk density is essential for soil management, and information about it is important in soil compaction as well as in the planning of modern farming techniques. Soil bulk density measurements are often required as an input parameter for models that predict soil processes. Such models often use bulk density measurements to account for horizon mass when aggregating soil data. Methods to measure bulk density are labor intensive and time-consuming. Thus, models have been developed to predict bulk density from soil physical and chemical data [1 – 3].

It includes all the elements (hydrogen, oxygen, nitrogen, etc) that are components of organic compounds, not just carbon. Soil organic carbon, soil organic matter and the correlation between bulk densities are frequently used to estimate carbon pools [4]. Organic matter plays an important Role for plants. Erdal Sakin [5] established the relationships between organic carbon, organic matter and bulk density in arid-semi arid soils in Southeast Anatolia region. Physico - chemical studies on varies soil have been widely studied [6 – 10].

The soil test aimed at soil fertility evaluation with resulting fertilizer recommendation is, therefore, the actual connecting link between the agronomic research and its practical application to the farmers' field. Soil testing can be divided into four steps (1) sampling (2) analysis (3) interpretation and (4) recommendations. One of the most important aspects of soil testing is that of obtaining a representative sample of the area.

### **Plant Nutrients:**

Although plants absorb a large number of elements, all of them are not essential for the growth of plant. The elements which are required by plant for their normal growth, development, metabolism and to complete their life cycle are called the essential ones. Some of these are required in large amounts and some in traces. Nutrients are classified as Primary (Macro), Secondary and micro, and are further classified as follow:

#### **Major nutrients required for plant growth**

Class 1: Carbon, hydrogen and oxygen (C,H, O).

Class 2: Nitrogen, potassium and phosphorus.

**Secondary Nutrients:** Magnesium, Calcium and sulphur

**Micro nutrients:** Iron, boron, zinc, molybdenum, manganese, copper and chlorine.

Soil testing can be divided into four steps (1) sampling (2) analysis (3) interpretation and (4) recommendations. One of the most important aspects of soil testing is that of obtaining a representative sample of the area.

### **CHEMICAL & EQUIPMENTS:**

Buffer tablate, Potassium Chloride, Potassium Dichromate, Sulphuric acid, Sodium bicarbonate, activated charcoal (phosphorous free), Ammonium Molybdate, Stannous chloride, Ammonium acetate, Calcium chloride, Glacial acetic acid, Barium Chloride, Gum acacia, Sodium diethyl dithiocarbamate, Sodium Hydroxide, Muroxide, Ethylene di amine tetraacetate, Ammonia buffer, Eriochrome Black-T, Diethylenetriamine pentaacetic acid were procured from s.d. fine chem Ltd. All chemicals are of analytical grade reagent.

pH was measured on pH meter (systronics Model No-335), Conductivity was measured on conductivity meter (systronics Model No-304), Optical density was measured on colorimeter (systronics Model No-202), Analytical balance (Wensar Model No-PGB200) was used to weigh samples and reagents, Flame photometer (systronics Model No-128) was used for analysis of Potash, Micro Nutrients was analyzed on Double beam atomic absorption spectrophotometer (Elico Model No-SL 194).

### **METHOD OF ANALYSIS:-**

#### **(1) Magnesium**

5 g air dried soil sample was taken in conical flask. To this, 25 ml of neutral ammonium acetate solution was added. The solution was shaken on mechanical shaker and filtered

through Whatman (No.1) filter paper. 5 ml solution was pipetted out in conical flask. To this solution, 2-3 crystal sodium diethyl dithiocarbamate, 5 ml of ammonium chloride-ammonium hydroxide buffer solution and 3-4 drops of Eriochrome black-T indicator were added. Titrated it slowly against 0.01 M EDTA solution. At the end point color changed from wine red to blue.

## **(2) Micronutrients (Cu, Fe, Mn, Zn) analysis by AAS**

### **Method for making D.T.P.A extracting solution**

1.967 g D.T.P.A. and 13.3 ml triethanol amine were taken in 500 ml flask. 400 ml distilled water was added. 1.47 g calcium chloride dihydrate was taken in 1ltr flask and dissolved in 400 ml distilled water. To this solution, D.T.P.A & T.E.A solution in calcium chloride dihydrate solution was added and adjusted pH 7.3 by using add 1M HCl. Make up 1 ltr with distilled water.

### **Method for micronutrients (Cu, Fe, Mn, Zn) analysis**

Weighted 20 g dried soil sample in a plastic bottle, then added 40 ml of D.P.T.A. solution. Shake on mechanical shaker for 2 hr. Filtered it on whatman filter No.40 in funnel cum test tube. Prepared standard curve for element by using different working ppm solution as per standard method of analysis and condition suggested by Elico brochure and then run the sample and note the ppm of elements. Obtained ppm reading multiplied with factor 2.0.

## **(3) Sulphur**

### **Method for making standard graph for Sulphur**

Weighted out 5.434 g potassium sulphate and make up 1 Ltr by using distilled water (this solution contains 1000 ppm of sulphur). 25 ml this solution was taken and make up 1 Ltr with distilled water (this is working standard solution of sulphur). Taken 0.0 (Blank), 1.0, 2.0, 3.0, 4.0, 5.0, 6.0, 7.0, 8.0, and 10 ml working solution in 25 ml volumetric flask. In every flask 1.0 g barium chloride and 1 ml gum acacia solution were added, and make up 25 ml by using distilled water. Then optical density of blank solution was set to zero using Blue filter.

**Table No-1 Reading For Standard Graph of Sulphur**

Sr. No.	Working standard sulphur solution in ml	Ppm	O.D
1	0	0	0
2	1	1	11
3	2	2	23
4	3	3	33
5	4	4	40
6	5	5	50
7	6	6	70
8	7	7	82
9	8	8	102
10	10	10	125
	Total ppm	46	Total: 536

**Calculation:-**

**1 Reading** = Total ppm of Sulphar/Total reading

$$1 \text{ Reading} = 46/536 \\ = 0.09$$

Sulphar ppm or mg/kg

$$\text{Sulphar ppm or mg/kg} = \text{sample reading} \times \text{graph Factor} \times 50 \times 25 / 20 \times 10$$

$$\text{Sample Reading} \times 0.084871 \times 50 \times 25/200$$

$$\text{Sulphar ppm} = \text{Sample Reading} \times 0.530443 \quad \text{or mg/kg}$$

**Process:**

10 g air dried soil sample was taken in 150 ml conical flask. 50 ml 0.15% calcium chloride extracting solution was added and shaken on mechanical shaker for 30 min. Filtered it on whatman filter No. 42. 20 ml filtrate was taken in 25 ml volumetric flask. 2 ml glacial acetic acid, 1 g crystal of barium chloride and 1 ml gum acacia solution were added. Make up the volume to 25 ml, then first set zero optical density using blank solution (as above method without taking soil sample) & then measure optical densities (O.D) of sample.

**(4) Calcium**

5 g air dried soil sample was taken in 150 ml conical flask and 25 ml of neutral 1 normal ammonium acetate was added. Shaken it on mechanical shaker for 5 min, and filtered through Whatman filter paper No.1. 10 ml filtrate solution was taken in conical flask, and 2-3 crystals of sodium diethyl dithiocarbamate were added. Then 5 ml 16% sodium hydroxide and 40-50 mg of the murexide indicator were added. Titrate it with 0.01N EDTA solution till the color gradually changes from orange red to reddish violet (purple), note the used EDTA solution.

**(5) Phosphorus**

**Method for making standard graph for phosphorus.**

0.439 g previously dried potassium dihydrogen orthophosphate was dissolved in 500 ml distilled water and 25 ml 7.0 N Sulphuric acid solution was added and then makes up 1 Ltr by using distilled water. 10 ml above solution was taken and makes up 500 ml by using distilled water (1 ml this resulting solution is equivalent to 2 ppm of phosphorus). By using this solution, various standard phosphorus ppm solutions were prepared and measured and their optical densities (O.D) were measured by using red filter.

**Table No-2 : Reading For Standard Graph of Phosphorus**

Flask No	2 ppm Working Solution of Phosphorous	8.5 pH Solution of Sodium Bicarbonate	1.5 Percentage Solution of Ammonium Molybdate-HCl	Working Solution of Steanus Chloride	O.D
1	0 Blank	5 ml	5 ml	1 ml	0
2	1 ml = 2 ppm	5 ml	5 ml	1 ml	26
3	2 ml = 4 ppm	5 ml	5 ml	1 ml	39
4	3 ml = 6 ppm	5 ml	5 ml	1 ml	55
5	4 ml = 8 ppm	5 ml	5 ml	1 ml	84
6	5 ml = 10 ppm	5 ml	5 ml	1 ml	104
7	10 ml = 20 ppm	5 ml	5 ml	1 ml	205
	Total = 50 ppm				513

### Calculation

1 Reading

$$= \text{Total Solution of ppm} / \text{Total Reading}$$

$$= 50 / 513$$

$$= 0.097$$

$$0.1010 \text{ Microgram P (Graph Factor)}$$

1 Gram Soil = R X 0.1010 X 4 Microgram P/ Gram Soil

R =Colorimeter Reading of Sample                      0. 1010= Graph Factor

P (Kg/ Hectare) = R X 0.1010 X 4 x 2.24                      (2.24 = Factor in 'P' Hectare)

P<sub>2</sub>O<sub>5</sub> Kg/ Hectare = R X 0.1010 X 4 x 2.24 X 2.29                      (2.29 = Factor in 'P<sub>2</sub>O<sub>5</sub> ' Hectare)

P<sub>2</sub>O<sub>5</sub> Kg/ Hectare = R X 2.0723584

### Process:

2 g soil sample and 40 ml 0.5 M sodium bicarbonate ( 8.5 pH) solution were taken in 100 ml beaker. To this, 1 g phosphate free activated charcoal was added and shaken on shaker for 30 minutes. The solution was filtered and pipette out 5 ml. 5ml 1.5% ammonium molybdate-hydrochloric acid solution was added to this solution. Allow to stand for 30 minutes, then 1ml 0.016 M stannous chloride solution was added make up 25 ml using distilled water. Blank solution was prepared according to the above process without taking the soil sample. Red filter was used and zero optical density was set by using above blank solution, then put the above sample solution and note the optical density.

## **(6) pH**

10 g soil & 20 ml distilled water were taken in 50 ml beaker & stirred for 30 min. Adjusted the temperature of pH meter at 25 °C. Calibrated the pH meter using 4, 7.0, 9.2 pH buffer solution. Washed the electrode by distilled water and clean by filter paper. Immersed electrode in above suspension solution and note the reading.

## **(7) Carbon**

### **Method for making standard graph for Organic carbon.**

Weighed out 1.25 g sucrose and taken it into 250 ml of volumetric flask and dissolved in 1 N of potassium dichromate solution, and makes up 250 ml volume by using 1 N potassium dichromate. 7 glass beakers of 50 ml were taken and numbered from 1 to 7. 0 ml, 1 ml, 2 ml, 3 ml, 4 ml, 5 ml and 6 ml solution was taken into above beakers from prepared solution of potassium dichromate. Take 10 ml 1 N potassium dichromate solution and 20 ml conc. sulphuric acid in test-tube and placed for 30 minutes. Allowed to cool and added 20 ml distilled water. Prepared following different standard carbon ppm solution and measured optical density (O.D.) by using red filter.

**Table No-3 : Reading for Standard Graph of Carbon**

Sr. No	ml of sucrose solution diluted in potassium dichromate	Amount of sucrose	O.D.
1	0 (blank)	-----	0
2	1	0.005 g	23
3	2	0.010 g	65
4	3	0.015 g	96
5	4	0.020 g	124
6	5	0.025 g	152
7	6	0.030 g	189
	Total	0.105 g	649

### **Calculation:-**

#### **1 Reading**

$$\begin{aligned} 1 \text{ Reading} &= \text{Total Amount of Sucrose} / \text{Total Reading} \\ &= 0.000161787 \\ &= 0.000161043 \text{ g Sucrose} \end{aligned}$$

1 Reading Carbon value:

$$\begin{aligned} &0.00006795 \\ &0.00006764 \text{ gram organic carbon} \end{aligned}$$

$$\begin{aligned} 1 \text{ Reading Graph Factor Value} &= 0.000067638 \times 100 \\ &= 0.0067638 \end{aligned}$$

**Process:**

1 g soil sample was taken in 100 ml beaker. 10 ml 1 N Potassium dichromate solution and 20 ml conc. Sulfuric acid were added to the sample and cooled the solution for 30 minutes. 20 ml distilled water was added and allowed for 12 hrs, then first set zero optical density using blank solution (as above method without taking soil sample & then measure optical density of sample.

**(8) Potassium**

**Method for graph factor of Potassium**

Prepared following stock solution and from it make various potash ppm solutions and run in flame photometer and note down potash ppm the reading.

**Table No-4 : Reading For Standard Graph of Potassium**

Flask No.	Stock solution	Concentration of Pottash in 100 ml Volumetric Solution (ppm)	Reading of Flame Photometer
1	0.0ml (Blank)	-----	0
2	1.0ml	10ppm	42
3	1.5ml	15ppm	45.5
4	2.0ml	20ppm	53.5
5	2.5ml	25ppm	61.5
6	3.0ml	30ppm	82
7	4.0ml	40ppm	106
	Total	140ppm	390.5

**Calculation**

$$\begin{aligned} 1 \text{ Reading} &= \text{Total Solution of ppm} / \text{Total Reading} \\ &= 140 / 390.5 \\ &= 0.359 \end{aligned}$$

$$\begin{aligned} 1 \text{ Gram Soil} &= R \times 0.359 \times 5 \text{ Microgram K} / \text{Gram Soil} \quad (0.359 \text{ Graph Factor}) \\ R &= \text{Flame Photometer Reading of sample} \end{aligned}$$

**K**

$$\text{Kg/Hectare} = R \times 0.359 \times 5 \times 2.24 \quad (2.24 = \text{Factor in K Hectare})$$

**K<sub>2</sub>O**

$$\begin{aligned} \text{Kg/Hectare} &= R \times 0.359 \times 5 \times 2.24 \times 1.20 \quad (1.20 = \text{Factor in K}_2\text{O Hectare}) \\ &= R \times 4.824 \end{aligned}$$

**Process:**

5 g soil sample was taken in 100 ml conical flask. 25 ml 1 M neutral ammonium acetate solution was added. Shaken it for 5 minutes on shaking machine and filtered the solution on whatman filter paper. Flame photometer was calibrated by using 10, 20, 30, 40, 50, 60, 70, 80 and 90 ppm standard potassium solution. After calibration, run above filtrate for analysis and note down the reading.

**(9) Electrical Conductivity (E.C.)**

10 g soil and 20 ml distilled water were taken in 50 ml beaker. It was stirred for 30 minutes. The temperature of E.C. meter was adjusted at 25 °C then conductance was adjusted to 1.412 mS/cm by using 0.01 N KCl solution. Washed the electrode with distilled water and cleaned with filter paper. Immerses electrode in above suspense solution and note the reading.

**Result and Discussion**

**Soil sampling**

Soil sampling was done during the dry season. Soil sampling was done at five randomly located points within each farm. The soils were sampled at two depths, 0 to 15 cm, 15 to 35 cm, using mini-soil pits dug at each sampling point. The soil samples were air dried in the laboratory and sieved through a 2 mm sieve for different types of laboratory analyses. The Results of soil samples & its LMH data shown in table no: 7(A), 7(B), 8(A), 8(B), 9(A), 9(B), 10(A), 10(B), 11(A), 11(B).



**Table No-5 : Critical Limits of Nutrients:-**

Sr.No	Parameters	Unit	Critical Limits		
			Low	Medium	High
1	pH	-----	<6.5	6.5-8.2	>8.2
2	Electric Conductance	-----	<1	1-3	>3
3	Organic carbon	%	<0.51	0.51-0.75	>0.75
4	Phosphorous	Kg/Hectare	<26	26-60	>60
5	Potash	Kg/Hectare	<151	151-300	>300
6	Zinc	ppm	<0.5	0.5-1.0	>1.0
7	Ferrous	ppm	<5	5-10	>10
8	Sulphur	ppm	<10	10-20	>20
9	Manganese	ppm	<5	5-10	>10
10	Copper	ppm	<0.2	0.2-0.4	>0.4
11	Magnesium	ppm	<1.0	1.0-2.0	>2.0
12	Calcium	ppm	<1.5	1.5-3.0	>3.0

**Calculation of soil fertility Index:**

$$= \frac{(\% \text{ of Low} \times 1) + (\% \text{ of Medium} \times 2) + (\% \text{ of High} \times 3)}{100}$$

**Table No-6 : Calculation of Low, Medium, High rating of soil fertility Index:**

Sr. No.	Range	Rating
1	Less than 1.67	Low
2	1.67 to 2.33	Medium
3	Greater than 2.33	High

**Table No-7(A) : Analysis of soil sample**

**Samples site :** Village : Jetalpur, Taluka: Radhanpur, District:Patan, Gujarat, India.

No	pH	E.C.	Org. Carbon (%)	Phosphorous (Kg/ Hectare)	Pottash (Kg/ Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me*/100 g soil)	Ca (Me*/100 g soil)
1	7.2	0.64	0.53	62.17	202.58	0.86	9.08	21.75	13.5	1.88	1.35	7.15
2	7.5	0.32	0.18	62.17	245.4	0.2	8.2	13.26	7.9	0.98	6.3	13.9
3	7.7	0.4	0.37	58.03	202.03	0.26	8.26	31.3	13.1	1.38	1.4	8.2
4	7.9	0.25	0.45	24.87	271.21	0.46	5.64	23.87	8	1.04	0.35	7.95
5	7.8	0.53	0.58	6.1	209.83	0.42	7.56	16.97	13.42	1.78	2.95	10.45
6	7.6	0.8	0.45	80.82	232.23	0.28	8.06	30.77	15.74	1.04	2.05	6.15
7	7.4	1.4	0.23	53.88	276.15	0.14	8.7	36.07	10.46	1.72	2.2	5.4
8	7.7	0.6	0.35	51.81	167.45	0.94	10.76	30.77	15.6	2.14	1.85	7.15
9	7.8	0.46	0.41	55.95	238.27	0.48	14.24	13.79	16.46	1.44	1.8	7.8
10	8	0.85	0.36	58.03	351.91	0.34	7.88	23.34	12.28	1.26	0.6	8.3
11	7.8	0.58	0.3	76.68	218.5	0.26	7.04	36.6	12.56	1.1	0.85	8.65
12	7.9	0.45	0.39	47.66	269.56	0.22	10.44	23.87	14.04	1.84	1.05	9.25
13	7.8	0.28	0.11	91.18	323.36	0.2	7.56	14.85	13.06	41.38	1.65	3.05
14	7.9	0.5	1.33	53.88	147.13	0.28	8.64	14.85	14.2	1.26	115	8.75
15	8.1	0.55	0.35	74.6	189.95	0.14	10.26	27.05	12.04	1.46	1.3	6.8
16	8.2	1.25	0.37	53.88	176.78	0.54	8.94	32.89	15.04	1.78	1.2	9.3
17	8	0.63	0.53	68.39	266.81	0.5	8.58	29.7	11.58	1.56	2.15	9.65
18	7.9	0.34	0.42	37.3	272.3	0.16	9.02	25.99	4.84	1.6	0.85	8.45
19	8	0.4	0.3	55.95	270.66	0.2	12.42	39.25	12.88	1.44	1.6	29
20	8	0.3	0.32	64.24	217.4	0.14	7.04	15.38	7.88	1.02	1	8.8

\*=Miliequivalent

**Table No-7(B) : Soil Fertility Index & Soil Test Rating:**

**Samples site :** Village : Jetalpur, Taluka: Radhanpur, District:Patan, Gujarat, India.

No	pH	E.C	Org. Carbon (%)	Phosphorous (Kg/Hectare)	Potash (Kg/Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me*/100 g soil)	Ca (Me*/100 g soil)
L	0	18	16	2	1	16	0	0	1	0	4	0
M	20	2	3	10	17	4	15	6	3	0	10	0
H	0	0	1	8	2	0	5	14	16	20	6	20
%L	0	90	80	10	5	80	0	0	5	0	20	0
%M	100	10	15	50	85	20	75	30	15	0	50	0
%H	0	0	5	40	10	0	25	70	80	100	30	100
S.F.I.*	2.00	1.10	1.25	2.30	2.05	1.20	2.25	2.70	2.75	3.00	2.10	3.00
LMH** of SFI	M	L	L	M	M	L	M	H	H	H	M	H

\*= Soil Fertility Index, \*\*= Low, Medium, and High Soil Fertility Index

**CONCLUSION:-** It is Concluded from above Analysis that E.C, Org. Carbon and Zn are in low amount , so Farm Yard Manure & Zinc Sulphate should be added for better plant growth & Productivity. The other parameters are sufficient. pH is in Medium limit so it can be Neutralized by using acidic fertilizer.

**Table No-8(A) : Analysis of soil sample**

**Samples site : Village : Maghapura, Taluka: Radhanpura, District: Patan, Gujarat, India.**

No	pH	E.C.	Org. Carbon (%)	Phosphorous (Kg/ Hectare)	Pottash (Kg/ Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me*/100 g soil)	Ca (Me*/100 g soil)
1	8.42	0.38	0.63	66.32	605.55	0.34	15.9	42.44	17.2	1.5	5.5	16
2	8.51	0.43	0.73	62.17	621.47	0.82	14.9	40.31	21.3	1.4	3.8	10
3	8.5	0.43	0.66	70.46	639.8	0.88	15.9	38.72	22	1.72	1.85	13.45
4	8.45	0.62	0.66	74.6	613.23	0.34	14.9	36.6	25.9	1.72	2.4	11.3
5	8.3	0.51	0.89	51.81	645.62	0.4	17	40.84	27	2.18	2.7	10.2
6	8.25	0.95	0.78	58.03	542.41	0.26	14	38.72	31.6	1.6	3.05	10.65
7	8.28	0.54	1.1	60.1	632.45	0.36	17.4	31.83	27.4	2.18	3.75	15.25
8	8.37	0.32	0.62	74.6	629.15	0.52	14	35.01	20.4	1.88	2.15	15.05
9	8.54	0.51	0.67	72.53	611.59	0.5	15.6	34.48	26.1	1.4	1.05	13.25
10	8.22	0.54	1.26	76.68	630.25	0.28	13.6	38.72	17.4	1.62	1.5	14.2
11	8.34	0.3	0.64	53.88	546.8	0.26	18.6	37.13	28.7	2.44	2.4	9.9
12	8.12	0.3	0.6	51.81	612.14	0.24	14.2	31.83	20.4	1.66	1.75	9.35
13	8.42	0.47	0.76	76.68	633	0.36	15.9	35.01	22.1	1.6	2.05	10.05
14	8.4	0.25	0.52	66.32	586.88	0.3	13.7	32.36	17.2	1.68	2.25	8.55
15	8.4	0.55	0.47	58.03	612.68	0.44	14.9	39.78	23.5	1.82	2.5	10.5
16	8.44	0.55	0.47	68.39	524.4	0.36	15.7	35.01	26.5	1.86	2.9	14.7
17	8.38	0.67	0.41	55.95	607.19	0.28	14	34.48	25.9	1.54	2.35	10.65
18	8.05	0.9	0.64	82.89	592.92	0.34	13.5	38.19	28.2	1.66	3.05	14.45
19	8.2	1.07	0.49	64.24	611.04	0.36	16	35.01	22.8	1.68	2.7	13.2
20	8.25	0.79	0.45	78.75	595.67	0.58	12.8	41.9	20.8	1.9	2.5	14.8

\*=Miliequivalent

**Table No-8(B) : Soil Fertility Index & Soil Test Rating**

**Samples site :** Village : Maghapura, Taluka: Radhanpura, District: Patan, Gujarat, India.

No	pH	E.C	Org. Carbon (%)	Phosphorous (Kg/ Hectare)	Pottash (Kg/ Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me*/ 100 g soil)	Ca (Me*/ 100 g soil)
L	0	19	5	0	0	15	0	0	0	0	0	0
M	3	1	10	6	0	5	0	0	0	0	4	0
H	17	0	5	14	20	0	20	20	20	20	16	20
%L	0	95	25	0	0	75	0	0	0	0	0	0
%M	15	5	50	30	0	25	0	0	0	0	20	0
%H	85	0	25	70	100	0	100	100	100	100	80	100
S.F.I.*	2.85	1.05	2.00	2.70	3.00	1.25	3.00	3.00	3.00	3.00	2.80	3.00
LMH** of SFI	H	L	M	H	H	L	H	H	H	H	H	H

\*= Soil Fertility Index, \*\*= Low, Medium, and High Soil Fertility Index

#### **CONCLUSION:-**

It is concluded from above Analysis that E.C & Zn are in low amount so Farm yard manure & Zinc Sulphate should be added for better plant growth & Productivity. The other parameters are sufficient. pH is in High limit so it can be Neutralized by using acidic fertilizer.

**Table No-9(A) : Analysis of soil sample**

**Samples site :** Village :Sherghadh, Taluka: Radhanpura, District: Patan, Gujarat, India.

No	pH	E.C.	Org. Carbon (%)	Phosphorous (Kg/ Hectare)	Pottash (Kg/ Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me*/100 g soil)	Ca (Me*/100 g soil)
1	8.5	1.51	0.93	68.39	321.71	0.5	5.64	35.01	3.1	1.34	3.4	16.2
2	7.95	0.5	0.6	62.17	398.03	0.3	4.54	31.83	5.08	0.62	2.55	10.15
3	8.8	1.52	0.62	53.88	361.24	0.32	5.94	38.72	6.26	1.02	2.3	13.2
4	8.13	1.31	0.64	47.66	383.2	0.26	3.8	39.25	6.34	0.7	2.7	12.9
5	7.95	1.4	0.61	49.74	372.77	0.4	4.9	18.04	5.9	1.16	2.35	17.15
6	7.9	1.8	0.24	66.32	202.03	0.2	3.94	22.81	4.06	0.52	2.2	6.2
7	7.92	1.53	0.14	51.81	143.29	0.12	4.38	13.79	2.32	0.4	2.05	7.15
8	8	1.47	0.45	62.17	268.46	0.16	5.78	15.38	3.04	0.74	2.3	9.2
9	7.91	1.37	0.6	58.03	411.2	0.26	4.54	27.58	2.8	0.4	2.75	10.05
10	7.85	1.39	0.57	72.53	464.45	0.2	4.32	21.22	2.88	0.44	1.65	9.55
11	7.9	1.47	0.45	68.39	322.81	0.14	4.9	21.75	3.58	0.62	0.4	11.4
12	7.95	1.62	0.51	66.32	420.53	0.72	4.54	32.89	5.24	0.4	0.75	11.85
13	7.92	1.56	0.47	55.95	333.79	0.3	5.64	31.83	4.58	0.78	0.65	13.65
14	7.82	1.39	0.59	29.01	346.42	0.3	4.02	39.78	4.5	0.26	1.4	15.1
15	8.05	1.06	0.26	64.24	255.29	0.24	3.3	39.25	3.78	0.74	1.1	11.4
16	7.9	1.44	0.44	68.39	404.06	0.28	4.32	22.81	4.02	0.66	1.7	11.5
17	7.8	1.37	0.45	37.3	313.48	0.2	4.02	30.24	3.86	0.44	1.15	13.15
18	8.17	1.36	0.37	41.45	263.52	0.16	3	15.91	3	0.26	3.45	9.05
19	7.91	1.48	0.55	43.52	246.5	0.42	4.38	14.32	3.58	0.56	0.75	12.45
20	8.09	1.71	0.49	51.81	394.73	0.18	4.76	18.04	3.32	0.44	0.9	13.9

\*=Miliequivalent

**Table No-9(B) : Soil Fertility Index & Soil Test Rating**

**Samples site :** Village :Sherghadh, Taluka: Radhanpura, District: Patan, Gujarat, India.

No	pH	E.C.	Org. Carbon (%)	Phosphorous (Kg/Hectare)	Pottash (Kg/Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me*/100 g soil)	Ca (Me*/100 g soil)
L	0	1	10	0	1	18	16	0	15	0	5	0
M	18	19	9	11	5	2	4	6	5	5	5	0
H	2	0	1	9	14	0	0	14	0	15	10	20
%L	0	5	50	0	5	90	80	0	75	0	25	0
%M	90	95	45	55	25	10	20	30	25	25	25	0
%H	10	0	5	45	70	0	0	70	0	75	50	100
S.F.I.*	2.10	1.95	1.55	2.45	2.65	1.10	1.20	2.70	1.25	2.75	2.25	3.00
LMH** of SFI	M	M	L	H	H	L	L	H	L	H	M	H

\*= Soil Fertility Index, \*\*= Low, Medium, and High Soil Fertility Index

#### CONCLUSION:-

It is concluded from above analysis that Org. Carbon, Zn, Mn and Fe are in low amount, so Farm Yard Manure, Zinc Sulphate, Ferrous ammonium Sulphate & Manganese Sulphate should be added for better plant growth & Productivity. The other parameters are sufficient. pH is in Medium limit so it can be Neutralized by using acidic fertilizer.

**Table No-10(A) : Analysis of soil sample**

**Samples site : Village :Arajansar, Taluka: Radhanpura, District: Patan, Gujarat, India.**

No	pH	E.C.	Org. Carbon (%)	Phosphorous (Kg/ Hectare)	Pottash (Kg/ Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me*/100 g soil)	Ca (Me*/100 g soil)
1	8.7	0.24	0.64	64.24	530.09	1.72	9.72	35.01	13.4	1.8	4.35	19.25
2	8.36	0.22	0.85	55.95	627.51	0.82	7.84	40.84	15.8	1.8	6.55	10.65
3	8.35	0.32	0.86	70.46	569.86	0.44	6.62	36.6	16.8	1.8	6.9	12
4	8.25	0.4	0.74	51.81	626.46	0.48	5.58	35.54	14.8	1.5	6.35	12.25
5	8.24	0.3	0.61	58.03	585.78	0.72	6.16	32.89	15.8	1	3.6	8.5
6	8.08	0.93	0.69	49.74	548.83	1.04	7	33.95	11.9	1.4	7.85	12.35
7	8.11	0.6	0.58	58.03	621.11	2.5	7.58	42.44	17.1	1.9	8.7	12.5
8	8.09	0.41	0.62	53.88	251.66	2.36	8.68	39.78	18.2	1.1	13.5	8.2
9	8.22	0.6	0.77	60.1	261.03	0.56	9.6	38.19	14	1.9	9.65	13.85
10	8.1	0.47	0.39	72.53	406.93	0.4	7.42	36.07	15.4	1	9.45	9.75
11	8.18	0.54	0.45	76.67	447.09	1.24	7.62	39.25	16.3	1.2	155	12.75
12	8.07	0.67	0.53	66.32	589.08	0.82	6.58	41.37	15.9	1.1	2.2	11.4
13	8.39	0.29	0.51	58.03	522.05	0.46	6.46	22.81	14.3	1.9	2.7	12.6
14	8.33	0.33	0.52	51.81	528.69	0.5	7.76	39.25	14.5	0.9	1.6	12.3
15	8.21	0.33	0.58	66.32	523.75	0.64	5.98	33.95	14.2	1.7	3.65	9.85
16	8.2	0.5	0.56	78.75	515.36	0.94	9.1	20.16	16.6	1	3.65	7.05
17	8.32	0.28	0.46	72.53	605	0.88	7.76	28.11	15.9	1.1	7.5	8
18	8.21	0.4	0.59	82.89	623.12	0.78	7.84	22.81	15.1	0.8	2.75	10.65
19	8.29	0.35	0.74	66.32	467.17	0.78	7	32.89	17.5	1.3	1	14.3
20	8.34	0.32	0.74	51.81	539.67	1	7.16	18.57	15	17	8.8	12.6

\*=Miliequivalent



**Table No-10(B) : Soil Fertility Index & Soil Test Rating**

**Samples site :** Village : Arajansar, Taluka: Radhanpura, District: Patan, Gujarat, India.

No	pH	E.C	Org. Carbon (%)	Phosphorous (Kg/ Hectare)	Pottash (Kg/ Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me*/ 100 g soil)	Ca (Me*/ 100 g soil)
L	0	20	3	0	0	4	0	0	0	0	0	0
M	7	0	14	9	2	11	20	1	0	0	2	0
H	13	0	3	11	18	5	0	19	20	20	18	20
%L	0	100	15	0	0	20	0	0	0	0	0	0
%M	35	0	70	45	10	55	100	5	0	0	10	0
%H	65	0	15	55	90	25	0	95	100	100	90	100
S.F.I.*	2.65	1.00	2.00	2.55	2.90	2.05	2.00	2.95	3.00	3.00	2.90	3.00
LMH** of SFI	H	L	M	H	H	M	M	H	H	H	H	H

\*= Soil Fertility Index, \*\*= Low, Medium, and High Soil Fertility Index

**CONCLUSION:-** It is Concluded from above Analysis that E.C. is Low Amount. The other parameters are sufficient. pH is in High limit so it can be Neutralized by using acidic fertilizer.

**Table No-11(A) : Analysis of soil sample**

**Samples site : Village : Dehgam, Taluka: Radhanpura, District: Patan, Gujarat, India.**

No	pH	E.C.	Org. Carbon (%)	Phosphorous (Kg/Hectare)	Pottash (Kg/Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me*/100 g soil)	Ca (Me*/100 g soil)
1	8.75	0.3	0.45	66.32	600.61	0.26	13.8	37.66	21.2	1.66	3	11.2
2	8.76	0.38	0.68	62.17	596.21	0.44	15.7	39.78	30.8	2.42	2.35	14.75
3	8.72	0.46	0.6	70.46	524.3	0.42	15.1	35.01	29.4	2.22	2.55	11.55
4	8.85	0.35	0.27	74.6	559.43	0.44	14.5	32.36	21	1.9	3.15	10.25
5	9.08	0.28	0.39	51.81	547.9	0.3	14	38.19	17.7	1.56	3.75	10.75
6	8.86	0.36	0.47	58.03	518.26	0.4	15.1	38.72	22.4	1.82	2	11.7
7	8.61	0.6	0.64	60.1	622.02	0.22	14.7	37.66	26.1	2.18	2.6	14.8
8	8.8	0.45	0.36	74.6	637.94	0.4	12.2	37.13	17.3	1.56	2.15	13.35
9	8.73	0.5	0.62	72.53	540.22	0.24	14.2	32.89	26.9	1.94	2.45	13.95
10	8.8	0.42	0.41	76.68	593.47	0.3	14	41.37	17.3	1.56	1.3	12.9
11	8.86	0.46	0.48	53.88	513.86	0.2	14.5	42.44	23.1	1.94	2.6	10.9
12	8.99	0.34	0.51	51.81	601.16	0.26	11.2	33.95	19.3	1.56	2.85	12.45
13	8.85	0.31	0.29	76.68	569.86	0.36	12.8	35.54	14.1	1.34	3.95	13.25
14	8.72	0.37	0.64	66.32	591.27	0.38	14.9	41.9	27.2	1.76	3.95	13.55
15	8.64	0.44	0.53	58.03	562.23	0.28	14.6	38.72	24.7	1.74	6.1	11.7
16	8.8	0.37	0.39	68.39	598.41	0.24	14.4	42.44	21.3	1.54	1.65	9.85
17	8.83	0.34	0.3	55.95	634.64	0.3	12.8	39.25	17.8	1.3	4.1	12.2
18	8.86	0.34	0.32	82.89	551.75	0.4	12.8	34.48	19.9	1.42	4.25	12.55
19	8.72	0.48	0.51	64.24	639.8	0.34	13.8	38.72	28.1	1.82	6.05	15.25
20	8.94	0.45	0.33	68.39	540.77	0.78	10.8	34.48	8.74	1.32	2.15	11.35

\*=Miliequivalent

**Table No-11(B) : Soil Fertility Index & Soil Test Rating**

**Samples site :** Village : Dehgam, Taluka: Radhanpura, District: Patan, Gujarat, India.

No	pH	E.C.	Org. Carbon (%)	Phosphorous (Kg/Hectare)	Pottash (Kg/Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me*/100 g soil)	Ca (Me*/100 g soil)
L	0	20	12	0	0	19	0	0	0	0	0	0
M	0	0	8	6	0	1	0	0	1	0	3	0
H	20	0	0	14	20	0	20	20	19	20	17	20
%L	0	100	60	0	0	95	0	0	0	0	0	0
%M	0	0	40	30	0	5	0	0	5	0	15	0
%H	100	0	0	70	100	0	100	100	95	100	85	100
S.F.I.*	3.00	1.00	1.40	2.70	3.00	1.05	3.00	3.00	2.95	3.00	2.85	3.00
LMH* * of SFI	H	L	L	H	H	L	H	H	H	H	H	H

\*= Soil Fertility Index, \*\*= Low, Medium, and High Soil Fertility Index

**CONCLUSION:-**

It is concluded from above Analysis that E.C., Zn and Org. Carbon are in low amount so Farm yard manure & Zinc Sulphate should be added for better plant growth & Productivity. The other parameters are sufficient. pH is in High limit so it can be Neutralized by using acidic fertilizer.

**REFERENCES:**

1. M. Bernoux, D. Arrouays, C. Cerri, B. Volkoff, and C. Jolivet, *Soil Sci. Soc. Am. Journal*, Vol.62, pp.743 – 749., 1998.
2. R.O. Curtis , B. W. Post , “Estimating Bulk Density from Organic Matter Content in Some Vermont Forest soils,” *Soil Sci. Soc. Am. Proc.*, Vol. 28, pp 285-286, 1964.
3. [31] E. Saki n, A. Deliboran and E. Tutar, “Bulk density of Harran plain soils in relation to other soil properties,” *African Journal of Agricultural Research*, Vol. 6(7), pp. 1750- 1757, 2011.
4. W.M. Post, W. R. Emmanuel, P. J. Zinke, A.G. Stangenberger, *Nature*, Vol. 298, pp. 156-159., 1982.
5. Erdal Sakin, *African Journal of Biotechnology* Vol. 11(6), pp. 1373-1377., 2012.
6. P. Qian, J. J. Schoenau and W. Z. Huang *Communications in Soil Science and Plant Analysis*, Vol. 23, No. 15-16, pp. 1791-1804, 1992.
7. S. Sato and N. B. Comerford, *Plant and Soil*, Vol. 279, No. 1-2, pp. 107-117, 2006.
8. T. Pare, E. G. Gregorich and B. H. Ellert, *Communications in Soil Science and Plant Analysis*, Vol. 26, No. 5-6, pp. 883-898, 1995.
9. M. B. Turrion, J. F. Gallardo and M. I. Gonzalez, *Communications in Soil Science and Plant Analysis*, Vol. 30, No. 7-8, pp.1137-1152, 1999.
10. P L Patel, Anita Gharekhan, N P Patel, P H Patel, *IMPACT: IJRANSS* Vol. 2, Issue 6,Jan,135-142, 2014.